# Lead Corrosion: What Research is Needed? Or:

"Aren't we done yet?"— The regulators' lament...



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# Lead and Copper Rule: US

- #First proposed: 1988
- #Covers all public water supplies and non-transient non-community supplies
  - -75,000+ total public water systems
  - -680+ over 50,000 population
  - -Administered at State level for 49 of 50
- **#Substantially revised and promulgated: 1991**

### Regulatory Approach

- #"Treatment Technique" rather than hard MCL for large systems
- **#Sampling scheme intentionally biased for site selection**

### Regulatory Approach

- #"Action Level" is trigger
  - Optimization of corrosion control (large)
  - Corrosion control studies and treatment to meet 0.015 mg/L for others
  - Public education
  - Possible service line replacement
- #Must meet other SDWA regulations at same time

## Origin of Continued Problems

- **#US** regulations primarily organized by "contaminant"
- # Pre-regulatory research not always completely comprehensive or accurate
- # Pre-regulatory research timeframe and resources often too limited
- **#** New regulations come into affect that interact.
- #So... Public water systems get to figure the rest out!

# Implementation Discoveries

- 1. Unexpected behavior after "corrosion control"
  - Different physical nature of plumbing or distribution systems than covered by research or that are amenable to the treatments
  - Different water chemistry conditions produce serious corrosion problems, e.g. Pb/Cu differences
- Thus, deficiencies in knowledge of lead corrosion are learned the "hard way!"

## Implementation Discoveries

- 2. PWS wish to make subsequent water quality changes
  - Changes in treatment caused by new regs
  - New water sources
  - Adverse secondary impacts
  - More stringent sludge/wastewater limits
- New questions arise to anticipate adverse health effects or water quality impacts

# Implementation Discoveries

- 3. Lag time of years may be needed for chemical reactions to take place
  - Water quality impacts (good or bad)
  - Hydraulics
- Utilities may only now be seeing effects of treatments started

vears ago

### **Objectives and Origins**

- #Based on both published and unpublished reports
- **#Questions involving** 
  - Significant data gaps
  - Reliable information lacking sufficient detail
- **#Spur new research initiatives**

### **Major Topics**

- **#Solder and Copper Alloy Corrosion**
- **#Lead Service Lines and Premise Plumbing**
- **#General Lead Chemistry and Toxicity**
- **#The Regulatory Balancing Act**

## Lead Solders and Copper Alloys

- #To what extent is control of lead release from brass fixtures and joints possible using only central water treatment?
  - -1-L first draw approx. 20-25 ft (6-8 m)
  - -Turbulent & erratic environment
  - -Any clear preferred strategy (e.g. pH-vs. ortho-P)?

## Lead Solders and Copper Alloys

#### **#Corollary Questions:**

- Are there optimizations that differ from those best for Pb pipe?
- -Will the problem go away with selective leaching, reduced Pb or reduced surface exposure?
- Does the timeframe differ with different waters?

## Lead Solders and Copper Alloys

**#Do cathodic inhibitors play a** significant role in limiting or mitigating lead release from these sources?

- **#Partial lead service line** replacement
  - -Elevated Pb levels?
  - -How high? How long?
  - –Relation to scale composition and/or water chemistry?
  - Can certain treatment strategies mitigate better?

- #To what extent are lead pipes covered with diffusion barriers or protective deposits composed of non-Pb solids?
  - -Al compounds
  - Mg silicates
  - Calcium and other carbonates or phosphates
- #How stable are these deposits in response to specific treatment changes?

#### **#Corollary questions:**

- –What are main elements or functional groups?
- To which water quality or treatment processes do they relate?
- –What is the optimum DIC/pH combination to produce most robust film in shortest time?
- If barrier films are unstable in new water chemistry, how long will it take for

- #Do the solubilities of basic lead carbonates and lead carbonate decrease with age?
  - Possible analogy to copper solids where metastability dominates
  - Would improve understanding and predictability of extrapolation from pilot scale or bench to field

- # Are there other significant solubilitycontrolling solids under certain water conditions (especially at lower pH's)?
  - -PbSO₄ [anglesite]
  - -Pb<sub>4</sub>SO<sub>4</sub>(CO<sub>3</sub>) [leadhillite/susannite/macphersonite]
  - -PbO·PbSO<sub>4</sub>
  - -Others?

- #How well can the best combination of startup inhibitor dosage and maintenance dosage be predicted?
- #What is the best startup approach for phosphate inhibitor dosage?
  - -High, then decrease
  - -Slowly increase

### General Lead Chemistry and Toxicity

- #When, if ever, are blended phosphates superior for controlling lead release?
- #Can the impact of source water NOM on lead release be predicted?
  - Most research results explain after the fact
  - Necessary to predict impact of D/DBP control strategies and simultaneous compliance

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### General Lead Chemistry and Toxicity

- #What are precise dosage/pH interrelationships for orthophosphate dosing across a wide range of water qualities?
- #How does the net lead solubility and reactivity of active inhibitor component vary with temperature or other "seasonal" factors?

### General Lead Chemistry and Toxicity

#Does the speciation of the lead resulting from different lead control strategies produce possibly different degrees of human bioavailability?

- #Can changes in coagulation type (e.g. alum to ferric chloride, or PACI) affect lead levels?
  - -Mechanism?
    - ! Scale solubility?
    - ! Destabilization by charge differences?

#What is the point of practical tradeoff between pH stability (buffer intensity) and possible increases in plumbosolvency or Pb release through added carbonate complexation?

- #To what extent does orthophosphate or polyphosphate(s) interact with residual aluminum?
  - Reduction of effectiveness of ortho-P for Pb or Cu control?
  - Formation of AI deposit reducing release
  - Adverse effect on hydraulics and aesthetics

#### **#Corollary questions:**

- Does a solid material form?
- Does the material have detrimental hydraulics effects?
- Which species are involved?
- Can the films be removed without detrimental effects on Pb or Cu?
- —If Al-based, does type of coagulant matter?

- #Are the products of chlorination or "advanced" oxidation of NOM more or less detrimental to lead release than "naturally-occurring" NOM species?
  - -Is O<sub>3</sub> without BAF detrimental?
  - Does the effect vary if pH/DIC is used as opposed to phosphate

### The Regulatory Balancing Act Fe/Mn interactions

- **#Do high redox potentials caused** by high DO levels (post O<sub>3</sub>) or Fe/Mn oxidation favor rapid passivation by PbO<sub>2</sub>?
- #What are the relative advantages and disadvantages of oxidation and physical removal vs. sequestration for different waters

- #What are the impacts of different types of phosphates on the passivation and lime leaching from cement pipes and linings?
  - -Phosphate chemical species effects
  - -Background water chemistry effects?

- #How important is overall Pb/Cu control optimization to levels beyond drinking water requirements to satisfy wastewater discharge and sludge limits?
- #Is more wastewater process research needed to optimize P, Zn, Cu, etc. removal?
- **#What are the impacts of different treatment approaches on hot water**

#### Conclusions

- # Considerable progress in 10 years, but
  - -Some old questions remain
  - New questions evolve out of other needs
- **#Bad timing for research funding when** in implementation stage
- #Lead control by central water treatment must be integrated into holistic scheme

## The Keys are in the Scales

- #Reflects past treatment and likely response to future changes
  - Mineralogy
  - Surface chemistry
- **#Overall, very few pipes have been studied**
- #Need to integrate knowledge of all distribution system materials, not just lead and copper